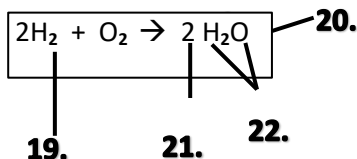
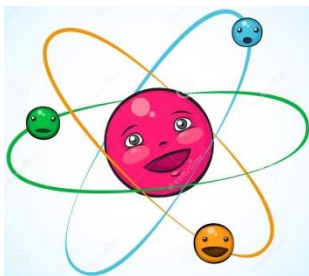


1. 2 or more substances mixed, but not chemically combined
2. Mixtures can be separated (chemically or physically)
3. List 3 examples of heterogeneous mixtures
4. Best mixed homogeneous mixture
5. Part of solution that does the dissolving
6. Part of solution that gets dissolved
7. A solution of 2 metals
8. Bronze is made from
9. Universal solvent
10. True / False- Mixtures are pure substances
11. To be able to dissolve in water
12. Cannot dissolve in water
13. Homogeneous mixture that is foggy
14. Raising the temperature, _____ solubility
15. Simplest of pure substances; cannot be changed
16. Smallest part of an element
17. When elements chemically combine, their properties . . .
18. Making sure there is the same number of atoms on both sides of an equation is called



23. To balance a chemical equation, change the
24. Total number of atoms? $2\text{Fe}(\text{NO}_3)_3$
25. Total number of atoms? $4\text{Ca}(\text{OH})_2$



1. _____
2. _____
3. _____
4. _____
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